

Dissolved oxygen profile in Ramsagar pond, Gaya

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Manuscript received online 30 July 2018, accepted on 10 November 2018

Abstract: Dissolved Oxygen is a potential Limnological dimension which reflects Physicochemical as well as biological health of ecosystem. Dissolved Oxygen controls metabolism of aerobic organisms present in aquatic ecosystem, it also affects the solubility. Dissolved Oxygen is contributed due to inputs from atmosphere and photosynthesis and output due to respiration, decomposition and mineralisation of organic matter and losses to the atmosphere. Dissolved Oxygen profile of Ramsagar Pond, Gaya was found to be 9.97 ± 0.255 with an inverse correlation with temperature of water and slight increase in monsoon months which may be caused due to additions of rain water in the pond.

(Keywords: Dissolved Oxygen, Metabolism, Winkler's Methods.).

Introduction

Dissolved Oxygen is essential to the metabolism of all aerobic organisms, oxygen distribution is important for the direct needs of many organisms, and effects the solubility and availability of aquatic ecosystem, Wetzel (1983). The items affecting the oxygen balance in water in water bodies are inputs due to the atmosphere and photosynthesis and output due to respiration, decomposition and mineralisation of organic matter and losses to atmosphere. Thus the oxygen balance in the water becomes poorer as the input of oxygen at the surface and photosynthesis activity decreases and as the metabolic performance of the heterotrophic organisms becomes greater. From this, a possibility of three situations may rise.

Experimental Material and Methods

Dissolved Oxygen a potential limnology dimension reflecting on physico chemical as well as biological health of aquatic systems, was calculated by Winkler's iodometric method² (1989)- modified by APHA³(1989), with the help of B. O. D bottle. B. O. D. bottle is a transparent glass bottle of known accurate volume (125cc-300cc corning) and is provided with a glass stopper.

The chemical used were sodium thiosulphate 0.025 N ($\text{Na}_2\text{S}_2\text{O}_3$), manganous sulphate ($\text{MnSO}_4 \cdot 4\text{H}_2\text{O}$), alkaline potassium iodide soln. (KI-KOH), sulphuric acid cone. (H_2SO_4 . Sp. Gr. 1.84) and starch solution. Efforts were made to complete the experiment at the earliest possible, so as to minimise the experimental errors.

Calculations was done by using following formula:-

$$\text{D. O. mg. L}^{-1} = (\text{ml} \times \text{N}) \text{ of titrant} \times 8 \times 1000 / V_2 (V_1 - V_3)$$

Where, V_1 = volume of sample bottle after placing the stopper.

V_2 = volume of the part of concentrated titrated.

V_3 = volume of MnSO_4 & KI added.

Observations

Table: Dissolved Oxygen(mg L⁻¹) profile of Ramsagar Pond during 2000.

Months	D.O.
January	7.42
February	7.10
March	5.96
April	6.26
May	4.90
June	4.55
July	9.51
August	14.40
September	11.34
October	6.36
November	6.44
December	7.59
Mean value with S. E.	9.97±0.255

Results and Discussion

Dissolved Oxygen (mg L⁻¹) was observed as minimum as 4.55 in June while as maximum as 14.40 in August with an annual average 9.97 ± 0.255.

In Ramsagar Pond the oxygen maxima has been, evidence in winter months and minima is

summer months except for monsoon months, where too, a state of maxima has been accounted for. This indicates a state of inverse relationship between DO and water temperature {Reid⁴, 1961; Welch⁵, 1952; Hussainy⁶, 1967; Gupta and Mehrotra⁷, 1986; Yadava et al⁸, 1987; {Charlton and Rao⁹, 1983}. This may probably be due to the higher rate of trophogenic activities at the surface water and oxygen uptake at the mud water interface. Yentsch and Ryther¹⁰ (1957) have suggested a declination in chlorophyll content for such lower values in summer times. Decomposition of organic matter may be an important factor in consumption of DO, which becomes more vigorous in warm weather {Gonzalves and Joshi¹¹, 1946, Begde & Verma, 1985}. Re-oxygenation of water in monsoon period may be due to circulation and mixing by inflow after monsoon rains {H.H Hannan et al. 1978; H.H Hannan et al, 1979; Bagde & Verma¹² 1985} with a progression of winter DO increased tremendously in Ramsagar Pond to its higher values, which may be due to circulation by cooling and draw down to dissolved oxygen in water {H.H Hannan¹³, Hannan et al¹⁴ 1979; Gordon & Nicholas¹⁵, 1970; Agrwal et al¹⁶. 1976; Rai¹⁷, 1978}.

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